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#### Key indicators

Single-crystal X-ray study T = 293 K Mean  $\sigma$ (C–C) = 0.008 Å R factor = 0.068 wR factor = 0.160 Data-to-parameter ratio = 10.8

For details of how these key indicators were automatically derived from the article, see http://journals.iucr.org/e. In the title compound,  $C_{13}H_{12}N_2OS_2$ , the whole molecule is nearly planar. The molecules are linked by  $N-H\cdots S$ hydrogen bonds between the amino group and the thione S atoms, forming molecular chains along the *a* direction. Received 18 April 2002 Accepted 7 May 2002 Online 17 May 2002

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## Comment

The study of dithiocarbazate derivatives and their metal complexes is prompted by their potential as biologically active agents (Tarafder, Saravanan *et al.*, 2001; Tarafder, Kasbollah *et al.*, 2001). The possible conformations formed in dithiocarbazate compounds and in their metal complexes are also of interest. Our attempts to prepare a new dithiocarbazate ligand led us to the title compound, (I). An X-ray structure analysis of (I) was undertaken in order to elucidate its conformation.



The bond lengths and angles in the title compound, (I) (Fig. 1), are within normal ranges (Allen *et al.*, 1987). The C11–N1, N1–N2 and C12–S2 bond lengths of 1.284 (6), 1.368 (5) and 1.665 (5) Å, respectively, are typical of Schiff base compounds of the dithiocarbazate system (S1/S2/C12/N2), *e.g.* S-methyl  $N^{\beta}$ -(4-nitrobenzal)methylenedithiocarbazate [the corresponding values are 1.265 (4), 1.376 (3) and 1.661 (2) Å, respectively; Duan *et al.*, 1997]. The title compound is in the *trans-cis* configuration, as was observed in other dithiocarbazate (Mattes & Weber, 1980) and S-benzyl dithiocarbazate (Shanmuga Sundara Raj *et al.*, 2000).



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The whole molecule of the title compound (Fig. 1) is nearly planar, with a dihedral angle of 3.1 (1)° between the dithiocarbazate (S1/S2/C12/N2) and naphthalene (C1–C10) systems. The planarity is due to the  $\pi$ -conjugation of the N1=C11 and N1–N2 joining the two systems. Atom C13 is displaced by 0.022 (2) Å from the dithiocarbazate plane, while the hydroxy atom O1 deviates by 0.025 (5) Å from the naphthalene plane.

In the title structure, there is an intramolecular O1– $H1A\cdots N1$  (Table 2) hydrogen bond, forming a six-membered N1–C11–C10–C1–O1–H1A ring. This interaction also contributes to the planarity of the title compound. The molecules are linked by three N–H···S hydrogen bonds (Table 2) between the amino group and thione S atom, forming a three-dimensional molecular network. Fig. 2 shows the packing diagram of (I), viewed down the *a* axis, showing the molecular chains along the *a* direction.

# **Experimental**

*S*-Methyldithiocarbazate (0.176 g, 0.01 mol) in ethanol was added to 20 ml of ethanol solution containing 2-hydroxynaphthaldehyde (0.172 g, 0.01 mol). The solution was stirred with a magnetic stirrer at 328 K for about 15 min. The resulting yellow precipitate was filtered and washed with ethanol followed by petroleum ether. Single crystals were grown by slow evaporation from a dimethyl sulfoxide solution.

## Crystal data

$\begin{array}{l} C_{13}H_{12}N_2OS_2\\ M_r = 276.37\\ Orthorhombic, P2_12_12_1\\ a = 5.1608 (1) Å\\ b = 9.4909 (2) Å\\ c = 25.8492 (7) Å\\ V = 1266.11 (5) Å^3\\ Z = 4\\ D_x = 1.450 \ {\rm Mg \ m^{-3}} \end{array}$	Mo $K\alpha$ radiation Cell parameters from 4198 reflections $\theta = 2.7-28.3^{\circ}$ $\mu = 0.41 \text{ mm}^{-1}$ T = 293 (2)  K Slab, yellow $0.50 \times 0.36 \times 0.18 \text{ mm}$
Data collection	
Siemens SMART CCD area- detector diffractometer $\omega$ scans Absorption correction: multi-scan ( <i>SADABS</i> ; Sheldrick, 1996) $T_{min} = 0.822, T_{max} = 0.930$ 7639 measured reflections	1819 independent reflections 1354 reflections with $I > 2\sigma(I)$ $R_{int} = 0.098$ $\theta_{max} = 28.4^{\circ}$ $h = -6 \rightarrow 6$ $k = -12 \rightarrow 9$ $l = -29 \rightarrow 34$
Refinement	
Refinement on $F^2$	H atoms treated by a mixture of

Refinement on  $F^2$   $R[F^2 > 2\sigma(F^2)] = 0.068$   $wR(F^2) = 0.160$  S = 1.011819 reflections 168 parameters

## Table 1

Selected interatomic distances (Å).

S1 C12	1 751 (5)	N1 C11	1 284 (6)
S1-C12 S1-C13	1.786 (6)	N1-N2	1.264 (0)
S2-C12	1.664 (5)	N2-C12	1.325 (6)

independent and constrained

-3

 $w = 1/[\sigma^2(F_o^2) + (0.0835P)^2]$ 

where  $P = (F_o^2 + 2F_c^2)/3$ 

refinement

 $(\Delta/\sigma)_{\rm max} < 0.001$ 

 $\Delta \rho_{\rm max} = 1.14 \text{ e} \text{ Å}$ 

 $\Delta \rho_{\rm min} = -0.66 \ {\rm e} \ {\rm \AA}^{-3}$ 



#### Figure 2

Packing diagram of the title compound, viewed down the *a* axis. The dashed lines denote the  $N-H \cdots S$  hydrogen bonds.

#### Table 2

Hydrogen-bonding geometry (Å, °).

$\begin{array}{c ccccccccccccccccccccccccccccccccccc$					
$\begin{array}{c ccccccccccccccccccccccccccccccccccc$	$D - H \cdots A$	D-H	$H \cdots A$	$D \cdots A$	$D - \mathbf{H} \cdot \cdot \cdot A$
	$\begin{array}{c} O1 - H1A \cdots N1 \\ C11 - H11 \cdots S2^{i} \\ N2 - H2A \cdots S2^{i} \\ C2 - H2 \cdots S1^{ii} \end{array}$	0.79 (6) 0.93 0.86 0.93	1.90 (5) 2.94 2.64 2.90	2.634 (5) 3.780 (5) 3.482 (4) 3.573 (6)	153 (6) 150 165 130

Symmetry codes: (i)  $\frac{1}{2} + x$ ,  $\frac{1}{2} - y$ , -z; (ii) 2 - x,  $\frac{1}{2} + y$ ,  $\frac{1}{2} - z$ .

The absolute configuration could not be determined, and Friedel pairs were merged. After location in a difference map, all H atoms, except the hydroxy H atom, were fixed geometrically and allowed to ride on the parent C or N atoms, with C-H = 0.97 Å and N-H = 0.89 Å.

Data collection: *SMART* (Siemens, 1996); cell refinement: *SAINT* (Siemens, 1996); data reduction: *SAINT*; program(s) used to solve structure: *SHELXTL* (Sheldrick, 1997); program(s) used to refine structure: *SHELXTL*; molecular graphics: *SHELXTL*; software used to prepare material for publication: *SHELXTL*, *PARST* (Nardelli, 1995) and *PLATON* (Spek, 1990).

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